

The modified expression for the Fourier transform after inclusion of the line density correction term becomes:

$$C(R, \psi, l/c) = \sum_m \sum_p \sum_d \sum_q \sum_s \rho_m \cdot J_p(2\pi R r_0) \cdot J_d(2\pi R \Delta) \cdot J_q(2\pi R \bar{r}_1) \cdot J_s(2\pi r_1 \sin \alpha \cdot l/c) E, \quad (3)$$

where

$$E = \exp[i(p(\psi - \varphi_0 + \frac{1}{2}\pi) + q(-\psi + \varphi_0 + \varphi_1 + \frac{1}{2}\pi) + d(\psi + \varphi_1 - \varphi_0 + \frac{1}{2}\pi) + s(\pi + \varphi_1) + 2\pi z_0 l/c),$$

subject to

$$N_0 p + (N_1 - N_0) q + (N_1 + N_0) d + N_1 s = l + m$$

where

$$q_0 = A' \cdot q_1 = q_{-1} = B'$$

and all  $q_i$  for  $|i| > 1$  are zero.

The theory has been checked by comparing results obtained from the expression of equation (3) with the optical diffraction pattern obtained from a mask which corresponded to a projection of a coiled-coil of constant line density and also, following a suggestion by Dr A. R. Stokes, by computing the Fourier transform obtained by a direct substitution of the coordinates of points  $(x_i, y_i, z_i)$ , spaced at small equal increments in  $\Delta s$  along the coiled-coil, into the expression

$$C(R, \psi, l/c) = \sum_{i=1}^N \exp[2\pi i(Xx_i + Yy_i + Zz_i)].$$

In both cases a good agreement was obtained (Pardon, 1966).

### Discussion

It has been shown that for models in which the original helix suffers appreciable distortion in forming a coiled-coil, the filament generated with the cartesian coordinate system adopted by Crick (1953) does not ap-

proximate to a filament of constant line density. The modified theory outlined above represents the Fourier transform of a model with much less variation in line density; this has been used in calculations for groups consisting of various numbers of coiled-coils packed coaxially. The theory has been found to be very satisfactory for models in which  $P=120 \text{ \AA}$ ,  $r_0=50 \text{ \AA}$ ,  $30 \text{ \AA} \leq r_1 \leq 40 \text{ \AA}$  and where  $N_0/N_1$  varied from 1 to 3.

If the original simple helix undergoes little distortion in forming the coiled-coil, as in the case considered by Crick (1953*a, b*), then the differences between the Fourier transforms calculated from the original theory and those from the modified theory are small. It seems probable that the original theory is sufficiently accurate for a model such as has been studied for  $\alpha$ -keratin.

The proposed super-helical model for nucleohistone and supporting experimental evidence has been recently described by Pardon, Wilkins & Richards (1967).

I should like to thank Professor M. H. F. Wilkins, Dr W. Fuller and Dr A. R. Stokes for helpful discussion, and to acknowledge the award of a Medical Research Council Scholarship.

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## The Structure of the $M'$ -Phase of $\text{YTao}_4$ , a Third Fergusonite Polymorph

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The  $M'$  phase of fergusonite is closely related to the other known monoclinic form ( $M$  phase), except that it has a halved  $b$  axis and its space group is  $P2/a$  instead of  $I2$ . The structure of the  $M'$  phase has been determined for  $\text{YTao}_4$  from three-dimensional diffractometer data, and its relationship to the other polymorphs is discussed. The lattice parameters of  $M'$ - $\text{YTao}_4$  are:  $a=5.292$ ,  $b=5.451$ ,  $c=5.110 \text{ \AA}$ ,  $\beta=96.44^\circ$ ,  $Z=2$ .

### Introduction

Using a natural fergusonite (impure  $\text{YNbO}_4$ ), Komkov (1959) determined the structure of the high-temperature form to be that of scheelite, space group  $I4_1/a$ ,

with oxygen coordinates close to those of  $\text{CaWO}_4$  (Kay, Frazer & Almodovar, 1964), and that of the low-temperature form to be a distorted scheelite, space group  $I2$ . The two forms, which will be referred to as the  $T$  phase and the  $M$  phase, respectively, are related by

a second-order (continuous) phase transformation (Gingerich & Bair, 1964); Fig. 1 illustrates how small the actual change is. Wolten & Chase (1967) give details of their discovery that the  $M$  phase can be obtained only as a transformation product of the  $T$  phase. If crystals are grown below the  $M$ - $T$  transformation temperature, a new monoclinic structure, the  $M'$  phase, is obtained. On heating, the  $M'$  phase is converted by a very sluggish first-order phase transformation to the tetragonal form which, when cooled, distorts slightly to give  $M$ .

Crystals of the  $M'$  phase have been grown for  $YTaO_4$  and for the lanthanon tantalates from Sm through Yb.  $YTaO_4$  was chosen for the structure determination reported here, because in this compound the difference between the scattering powers of the two kinds of metal atom is a maximum. The objective of the work was to ascertain the structure type of the new polymorph and to compare it with the other two known forms.

### Experimental

Table 1 gives the lattice parameters of  $YTaO_4$  in its two monoclinic forms. Those for the  $M'$  phase were determined from single-crystal diffractometer data under high-resolution conditions, using copper radiation,  $K\alpha_1 = 1.54051 \text{ \AA}$ . Compared with the  $M$  phase, the newly discovered  $M'$  phase, apart from belonging to a different space group, has a halved  $b$  axis and  $a$ ,  $c$ , and  $\beta$  parameters that are about 1% larger.

Table 1. Lattice parameters of monoclinic  $YTaO_4$

| Space group | $M'$ phase<br>(this work)<br>$P2_1/a$ | $M$ phase<br>(Keller, 1962)<br>$I2$ |
|-------------|---------------------------------------|-------------------------------------|
| $a_0$       | 5.292 $\text{\AA}$                    | 5.239 $\text{\AA}$                  |
| $b_0$       | 5.451                                 | 10.893                              |
| $c_0$       | 5.110                                 | 5.056                               |
| $\beta$     | 96.44°                                | 95.52°                              |
| $Z$         | 2                                     | 4                                   |

All single crystals of the  $M'$  phase are thin (010) laths, elongated parallel to  $a_0$ . No equant crystals were found, and the crystals could not be shaped. The one chosen for this investigation had the typical dimensions  $0.330 \times 0.115 \times 0.023 \text{ mm}$ . The absorption coefficient of  $YTaO_4$  is  $962.23 \text{ cm}^{-1}$  for Cu radiation and  $319.5 \text{ cm}^{-1}$  for Ag radiation. It was therefore essential to calculate absorption corrections accurately and to do this from the actual shape of the crystal rather than an approximation. This was accomplished with the aid of a recent version of the computer program ORABS (Wehe, Busing & Levy, 1962). The limiting factor in these calculations is the precision with which one can measure and describe to the program the shape and size of the crystal. To minimize this source of error, a low absorption coefficient is desirable, leading to the choice of Ag radiation,  $K\alpha_{av} = 0.5608 \text{ \AA}$ . The transmission factors calculated for the crystal employed in this work covered a range of 40:1, from a high of 0.1859 for the 091 reflection to a low of 0.0047 for the  $\bar{2}01$  reflection.

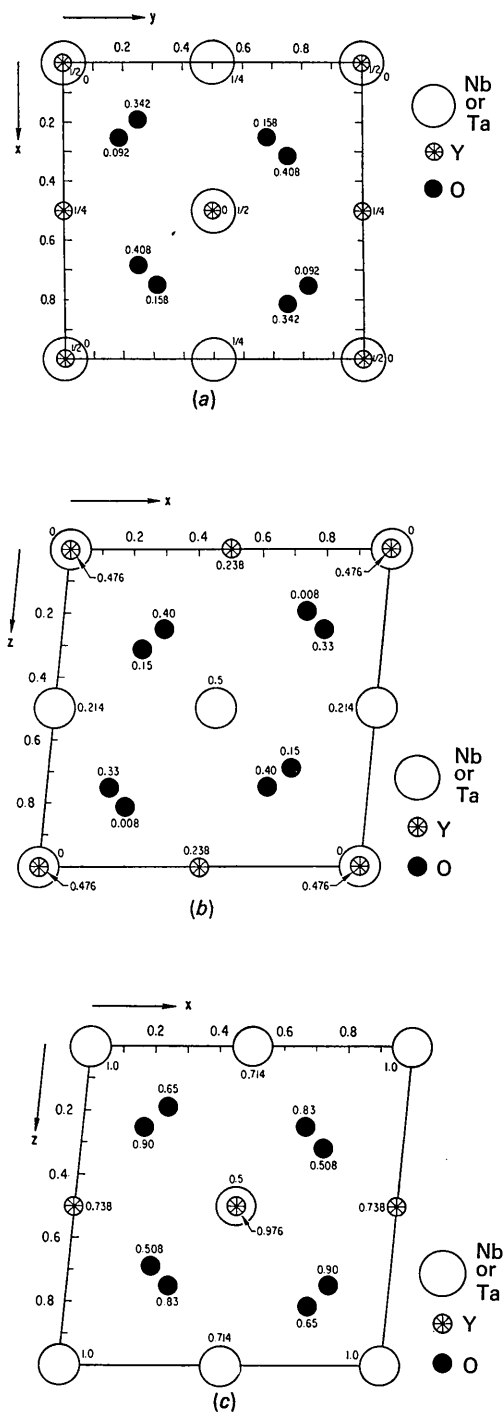


Fig. 1. Comparison of  $M$  and  $T$  fergusonite structures. (a) Tetragonal fergusonite: unique( $c$ )-axis projection of tetragonal (scheelite-type)  $YTaO_4$ . One half of the unit cell ( $z$  from 0 to 0.51) is shown; the other half is rotated  $90^\circ$  about  $c$ . Data from Komkov. (b) Monoclinic fergusonite,  $M$  Type: unique( $b$ )-axis projection of one half ( $y$  from 0 to 0.5) of  $M$ -type fergusonite cell. Data from Komkov  $00\bar{1}/010/100$ . (c) Monoclinic fergusonite,  $M$  type: unique( $b$ )-axis projection of upper half ( $y$  from 0.5 to 1.0) of  $M$ -type fergusonite cell.

The magnitude of the absorption corrections makes them a major factor in the conversion of observed intensities to structure factors. Since they have been calculated from dimensional measurements some of which are only accurate to two significant figures, it is obvious that they limit the possible accuracy of the structure factors. The final list of  $F_o$  versus  $F_c$  shows a correlation between large absorption corrections and large values of  $F_o-F_c$ .

The reciprocal sphere for Ag radiation contains over 5000 independent reflections for this crystal. An automated diffractometer not being available, reflections were measured only at values of  $2\theta < 60^\circ$ , except for nine additional  $0k0$  reflections to  $2\theta = 135^\circ$ . Omitting space group absences and a few of the strongest reflections suspected of extinction, 729 reflections, including 16 unobserved ones, entered into the refinement. For the first 100 reflections, both  $hkl$  and  $h\bar{k}l$  were measured and averaged. The mean reproducibility was 2% for strong and 5% for weak reflections.

The intensities were measured by means of  $\theta-2\theta$  scans; backgrounds were measured by stationary counts on the low and high angle sides of each reflections. The detector was a scintillation counter. Two standard reflections were measured at frequent intervals to check for variations of the primary beam intensity, and scale factors were applied when indicated.

Since the diffractometer was operated manually, each reflection was first observed on the strip-chart recorder to determine an appropriate scanning range for the intensity count. Both components of the  $K\alpha$  doublets were always included.

### The structure determination

From qualitative intensity considerations it was clear that the yttrium and tantalum atoms could be placed only in equipoints  $2(e)$  and  $2(f)$ , with coordinates  $\frac{1}{4}, y, 0$ ;  $\frac{3}{4}, \bar{y}, 0$  and  $\frac{1}{4}, y', \frac{1}{2}$ ;  $\frac{3}{4}, \bar{y}', \frac{1}{2}$ . Several possible values of the two  $y$  parameters were derived from selected reflections, and a particular combination of two of these was confirmed by two appropriate Patterson sections. The coordinates of the metal atoms were then used in a three-dimensional Fourier synthesis which located the oxygen atoms.

The coordinates were refined by full-matrix least squares, using anisotropic temperature factors for Y

and Ta and isotropic ones for oxygen. The final  $R$  value was 0.149, which is considerably higher than the reproducibility of the intensity measurements had led us to believe.

Absorption errors, already discussed, may have contributed to the high value of  $R$ . In addition, inaccuracies in the data may have arisen from one or both of the following factors.

1. There may have been some disorder in the crystal. This point was also raised by a referee who suggested substitutional disorder. Substitutional disorder is quite unlikely in this compound, but the existence of two other polymorphs suggests the possibility of random displacements of some atoms towards positions they would occupy in one of the other polymorphic modifications. Such displacements would occur mostly – if indeed they do – in the  $xz$  plane, and the final list of  $F_o$  versus  $F_c$  shows the worst fits for  $h0l$  reflections. Our first batch of crystals, before we learned to grow them better, actually had some extra X-ray reflections which were tentatively interpreted as due to nuclei of the tetragonal phase upon which the crystals of the  $M'$  phase had grown. Even later batches, which furnished the crystal used here, were never perfect enough to give recognizable interference figures in the conoscop. It is true that all the crystals were quite small, but a really good crystal will give a recognizable optic figure even if very small. These observations lend a measure of support to the suggestion of some degree of disorder.

2. While a silver target X-ray tube will give the optimum ratio of characteristic to continuous radiation at potentials from 75 to 90 kV, it had to be operated at 50 kV because of equipment limitations. At the same time, the recommended filter thickness (Rh) of 3.5 mils was reduced empirically to about 2 mils to obtain reasonable intensities. While no  $\beta$  lines were observable, the monochromaticity of the beam may not have been as good as desirable. A number of reflections had pronounced 'tails', and while these may be interpreted as due to disorder, they are more likely – in retrospect – to be due to background radiation, corresponding to the general radiation streaks visible on precession photographs. It may well be that in a number of cases the existence of these 'tails' caused errors in apportioning the intensity properly between reflection and background.

Table 2. Atomic coordinates and temperature factors

The numbers in parentheses are the standard deviations in the last digits given.

| Atom Equipoint | Yttrium 2(e)  | Tantalum 2(f) | Oxygen 1 4(g) | Oxygen 2 4(g) |
|----------------|---------------|---------------|---------------|---------------|
| $x$            | $\frac{1}{4}$ | $\frac{1}{4}$ | 0.490 (6)     | 0.081 (12)    |
| $y$            | 0.7694 (6)    | 0.3056 (3)    | 0.462 (4)     | 0.069 (8)     |
| $z$            | 0             | $\frac{1}{2}$ | 0.266 (4)     | 0.258 (7)     |
| $B$            | —             | —             | 0.59 (30)     | 3.12 (75)     |
| $B_{11}$       | 0.0024 (108)  | 0.0024 (6)    | —             | —             |
| $B_{22}$       | 0.0033 (7)    | 0.0047 (4)    | —             | —             |
| $B_{33}$       | 0.0049 (12)   | 0.0068 (6)    | —             | —             |
| $B_{13}$       | -0.001 (2)    | 0.0064 (10)   | —             | —             |

Results

It has been noted that much of the responsibility for the relatively high R value rests with reflections of the type h0l. Since the yttrium and tantalum atoms are in special positions with only the y parameter variable, it is believed that they have been determined quite well. The oxygen positions may not be better than 0.05 Å. This is sufficient to establish the nature of the coordination polyhedra but not the degree of distortion of the latter from cubes and tetrahedra respectively.

Most of the temperature factors have reasonable values but a few do not. With an R value barely below 15%, it is questionable how much importance can be attached to their exact numerical values, hence they are quoted without further comment.

The positional parameters, temperature factors and standard deviations derived from the least squares procedures are listed in Table 2. Fig. 2 is a b-axis projection of the electron density, Table 3 gives some distances and angles, and Table 4 is the list of F<sub>o</sub> versus F<sub>c</sub>.

Table 3. Distances and angles\*

The first column lists all the distinct metal-oxygen distances, the second column the angle subtended at the metal atom by the equidistant oxygen pair having the M-O separation to the left.

One oxygen atom in the yttrium coordination sphere, the one at a distance of 2.40 Å, approaches a tantalum atom to within 2.14 Å.

Table with 2 columns: Y-O and O-Y-O distances and angles. Y-O values: 2.25 Å, 2.34, 2.40, 2.42. O-Y-O values: 113.6°, 71.5, 95.2, 76.4. Ta-O values: 1.94 Å, 2.03. O-Ta-O values: 74.6°, 105.0.

\* As is to be expected from the similarity of the lattice parameters, the Y-O distances are almost identical with those in the M phase, which range from 2.26 to 2.45 according to Komkov. In Y<sub>2</sub>O<sub>3</sub>, where the oxygen cubes are less distorted, the Y-O distances are 2.26 to 2.354 (Paton & Maslen, 1965). The Ta-O distances in NaTaO<sub>3</sub> and KTaO<sub>3</sub> are 1.93 and 1.99 respectively (Vousden, 1951). This would seem to justify the conclusion that the metal-oxygen distances are more accurate than the oxygen positional parameters are estimated to be.

Computational details

Atomic scattering factors were taken from International Tables for X-ray Crystallography (1962). The quantity minimized by least squares was the sum of the weighted and squared differences between F<sub>o</sub><sup>2</sup> and F<sub>c</sub><sup>2</sup>, with weights essentially inversely proportional to the observed intensities. The equations are given in the CRYRM manual (Duchamp, 1964).

During the final cycle of least squares, all of the positional parameters shifted by about 0.2 standard deviations, except x for O(2) which moved 0.9 s.d. All

temperature factors shifted between 0.01 and 0.05 s.d., except B<sub>13</sub> for Ta which changed by 0.09 and B for O(2) which changed by 0.08 s.d.

Discussion of the structure and phase transformation

The yttrium atoms are surrounded by distorted cubes of eight oxygen atoms, and the tantalum atoms by distorted tetrahedra of four oxygen atoms. Because of

Table 4. Observed and calculated structure factors

Table with multiple columns (h, k, l, I<sub>o</sub>, I<sub>c</sub>) containing observed and calculated structure factors for various Miller indices. The table is organized into several sections corresponding to different reflections.

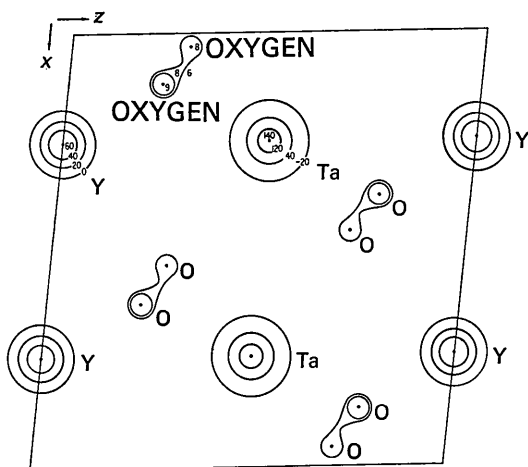


Fig. 2. YTaO<sub>4</sub>, *M'* phase: *b*-axis projection of electron density. Contour lines labelled in e.Å<sup>-2</sup>.

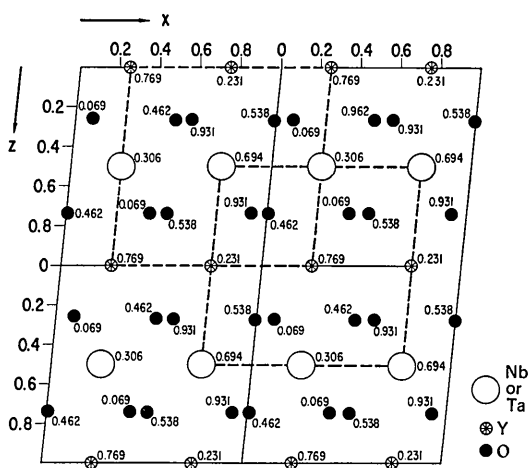


Fig. 3. Monoclinic fergusonite, *M'* type: unique(*b*)-axis projection of full cell. Four unit cells are shown side by side. Dashed cells have origins shifted in different ways to correspond to bottom and top halves of *M*-type or scheelite cells. Transformation involves shift of every second cell along *b* through  $a/2$ ,  $c/2$  and adjustment of oxygen atoms.

the location of the metal atoms on the twofold axes, all Y–O and Ta–O distances occur in pairs.

The drawing of Fig. 3 is a *b*-axis projection of four unit cells of the structure, based on the results of the least-squares refinement. The parameters derived from the Fourier synthesis differ by amounts ranging from 0.01 to 0.03, which is the right magnitude for series termination errors; hence, they have been so interpreted. This is the reason why two of the oxygen atoms which in the drawing appear on a line almost parallel ( $\Delta x=0.01$ ) to the *z* axis are considerably more inclined to the *z* axis in the Fourier synthesis ( $\Delta x=0.07$ ). The discrepancies in the *x* coordinates, assumed to be termination errors, are in opposite senses.

The dashed outlines indicate alternate unit cells with the origins translated in such a manner as to agree with the drawings of the *T*- and *M*-structures shown in Fig. 1. The two dashed cells are displaced with respect to one another by  $a/2$ ,  $c/2$ . If they are brought into register, then one cell of the *M* structure is obtained, except for twisting the groups of oxygen atoms. Individual oxygen atoms, however, move only through small fractions of interatomic distances.

If the formal mechanism of the *M'*-to-*M* transformation just described, namely displacement of alternate unit cell through  $a/2$ ,  $c/2$  and adjustment of oxygen positions, is also the actual mechanism, then this is a very interesting transformation.

In the sense that there is no diffusion of individual atoms, but cooperative motion of groups of atoms, there is a startling resemblance to diffusionless shear transformations, as in ZrO<sub>2</sub> (Wolten, 1963, 1964). But unlike such transformations, not all the atoms move in unison, and the transformation is definitely not martensitic. Small groups of atoms, the building blocks of the structure, are sheared along certain planes and move as units through distances that exceed interatomic distances; this action would be analogous to the manner in which individual atoms diffuse in more conventional ('nucleation and growth' type) phase transformations. Since many strong bonds are broken in the process, the sluggish nature of the transforma-

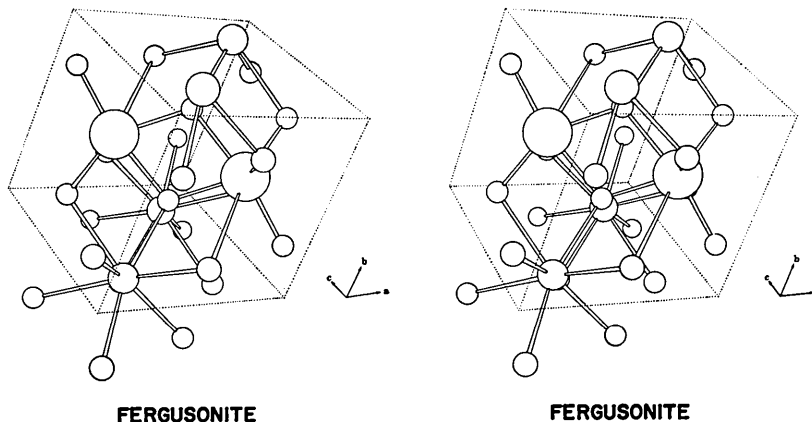


Fig. 4 Stereoscopic drawing of the structure of *M'*-type fergusonite.

tion is readily understood. Of course, in the absence of compelling evidence, translation of such large blocks *in toto* must be considered unlikely. The conservative view is to postulate the reorganization of the structure to be brought about by the diffusion of much smaller blocks, consisting of from one to perhaps three atoms each, which then reassemble in a pattern that recreates many of the features of the original structure.

A stereoscopic drawing of the structure of the  $M'$  phase is shown in Fig. 4.

The Patterson, Fourier, and least-squares calculations were performed at the California Institute of Technology under the CRYRM crystallographic computing system (Duchamp, 1964) through the generosity of Prof. R. E. Marsh, to whom the author is also greatly indebted for his expert advice on data handling. Dr W. R. Busing, Oak Ridge National Laboratory, was kind enough to furnish the computer program ORABS. Thanks are also due Mr W. Gehrler who patiently, over a period of many months, measured the intensities and backgrounds one by one.

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## The Crystal Structure of a Double Oxalate of Yttrium and Ammonium, $NH_4Y(C_2O_4)_2 \cdot H_2O$

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A three-dimensional X-ray analysis of  $NH_4Y(C_2O_4)_2 \cdot H_2O$  has yielded the positions of all the atoms except hydrogen with standard deviations of  $0.0012 \text{ \AA}$  for Y, and less than  $0.02 \text{ \AA}$  for the other atoms. The crystals are monoclinic with  $a = 9.18 \pm 0.01$ ,  $b = 6.09 \pm 0.01$ ,  $c = 7.89 \pm 0.01 \text{ \AA}$ ,  $\beta = 90.2^\circ \pm 0.1^\circ$ ; space group  $P2_1/n$ ;  $Z = 2$ . There is ninefold coordination; eight oxygen atoms and the  $H_2O$  form a slightly distorted trigonal prism arrangement around the yttrium, at distances ranging from  $2.343 \text{ \AA}$  (for the  $H_2O$ ) to  $2.412 \text{ \AA}$ . The two independent oxalate ions are planar and have crystallographic centres of symmetry. Their dimensions are in agreement with the results of previous work on oxalates.

### Introduction

Hydrated oxalates of the rare earths having the composition  $Ln_2(C_2O_4)_3 \cdot 10H_2O$  ( $Ln = \text{lanthanide}$ ) have been known for some time, and are thought to be interstitial hydrates though their detailed structures are not known (Gilpin & McCrone, 1952). Recently, Barrett, McDonald & Topp (1964) made a study of the precipitation of the rare earth oxalates from oxalic acid solution, in the course of which a new series of double salts was discovered with the composition  $NH_4Ln(C_2O_4)_2 \cdot nH_2O$ . Yttrium and the elements in the group Sm–Tm formed monohydrates which appeared to be isomorphous, while the remaining rare earths formed trihydrates exhibiting a number of different

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X-ray powder diffraction patterns. The monohydrated yttrium salt was obtained as single crystals and the present paper describes an X-ray analysis of the structure. The other members of the monohydrate series presumably also have this structure.

### Experimental

$NH_4Y(C_2O_4)_2 \cdot H_2O$  is precipitated from strongly acidic solutions containing an excess of oxalate ion by the addition of ammonia. Prepared in this way, it is a fine powder. Crystals for the present work were grown hydrothermally. The prisms so obtained showed development of the forms  $\{010\}$  and  $\{101\}$  and were invariably twinned.

The intensity data were collected from three sets of equi-inclination Weissenberg photographs:  $(0-5, k, l)$ ,  $(h, k, 0-6)$ , and  $(h, 0-4, l)$ . Nickel-filtered  $Cu K\alpha$  radi-

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